Appendix 6: Electronegativies (BLM)

Use the table of electronegativities to determine the bond type (ionic, polar covalent, non-polar covalent) that would be formed between each of the following elements. Provide the electronegativity difference for each pair.

		Bond Type	Electronegativity Difference
1.	Na, Cl		
2.	Al, Cl		
3.	H, S		
4.	K, F		
5.	O, O		
6.	Mg, S		
7.	Li, Br		
8	FF		